S.N.D.T. WOMEN'S UNIVERSITY



Proposed Syllabus for

M.Sc. Programme

Subject: Analytical Chemistry

Revised-2014

Eligibility: B. Sc. with 5 units of 4 credits each (Annual System) or 32 credits (Semester System) of Chemistry.

FACULTY : Science

TOTAL CREDIT : 96

COURSE : Masters in Analytical Chemistry (M.Sc.)

SEMESTER I

					Total	
Paper	Title	Hrs/wk	Internal	External	Marks	Credit
1001	Fundamentals of Analytical Chemistry	4	50	50	100	4
1002	Food & Biochemical Analysis	4	50	50	100	4
1003	Environmental Science	4	50	50	100	4
1004	Drug Laws and Packaging	4	50	50	100	4
1005	Practical Analytical Chemistry	8	50	50	100	4
1006	Practical Food and Biochemical Analysis	8	50	50	100	4
				Total	600	24

SEMESTER II

					Total	
Paper	Title	Hrs/wk	Internal	External	Marks	Credit
2001	Electro Analytical & Spectroscopic Methods	4	50	50	100	4
2002	Pharmaceutical Analysis	4	50	50	100	4
2003	Cosmetics Formulations & QC	4	50	50	100	4
2004	Research Methodology	4	50	50	100	4
2005	Practical Spectroscopy and Chromatography	8	50	50	100	4
2006	Practical Pharmaceutical Analysis	8	50	50	100	4
				Total	600	24

SEMESTER III

					Total	
Paper	Title	Hrs/wk	Internal	External	Marks	Credit
	Advanced Chromatography & Spectroscopic					
3001	Methods	4	50	50	100	4
3002	Organic Analysis	4	50	50	100	4
3003	Microbiological Methods of Analysis	4	50	50	100	4

3004	Elective Paper	4	50	50	100	4
3005	Practical Advanced Spectroscopic Methods	8	50	50	100	4
3006	Practical Organic Analysis	8	50	50	100	4
				Total	600	24

SEMESTER IV

					Total	
Paper	Title	Hrs/wk	Internal	External	Marks	Credit
4001	Advanced Analytical Techniques (9 wks)	4	50	50	100	4
4002	Practical Advanced Anal. Techniques (9 wks)	8	50	50	100	4
4003	Research Project (9 wks)	12	100	100	200	8
4004	In-plant training (6 weeks)	40	100	100	200	8
				Total	600	24

25 Marks, 1 credit, 15Hrs (1 lecture=1 hour)

Elective

- 1. Industrial Products & Forensic Analysis
- 2. Medicinal Chemistry
- 3. Biosensors, Agrochemicals & Organic Polymers

Methodology

- 1. Lectures, Tutorials
- 2. Self study/ Internet/ Websites
- 3. Home assignments

Internal Assessment (Any One)

1. Presentation on one of the topic from the

text

- 2. Community extension work
- 3. Field survey

	SEMESTER I				
Code: 1001	Title: Fundamentals of Analytical Chemistry	Credits: 4			
Objectives: I)	To understand the basic concepts of analytical techniques.				
ii) Able to prep	pare standard solutions.				
iii) To acquire knowledge of theoretical concepts of volumetric techniques.					
iv) To develop	expertise in collection, preparation and preservation of samples.				
v) Participants	will use statistical aids to compile, tabulate, evaluate and present a	nalytical data.			
Paper I		60 Hours			
Unit I	Analytical Basics:	15 Lectures			
	a) Analytical chemistry - Scope, functions and analytical process.				
	b) Sampling: Collection, Preservation and preparation of				
	sample,Techniques of sampling solids, liquids and gases;				
	Operation of drying and preparing a solution of the analyte.				
	c) Chemometrics				
	Topics to be covered in the form of numerical				
	problems:Concentration of a solution based on volume and mass				
	units, Calculation of ppm, ppb and dilution of the solutions,				
	Concept of mmol and kgmol, Stoichiometry of chemical				
	reactions, Limiting reactant, Theoretical and practical				
	yield,Calculation of pH of acids,bases and acidic and basic				
	buffers; Concept of formation constant, Stability and instability				
	constant,Stepwise formation constants.				
Unit II	Volumetric Methods of Analysis	15 Lectures			
	Introduction to Volumetric Methods:				
	Calibration of volumetric apparatus, Primaryand Secondary				
	Standards, Principles of volumetric analysis, Acid-base titration.				
	Titration in non-aqueous solvents, Complexometric titrations,				
	Precipitation titrations, Redox titrations, Theoretical aspects of				
	titration curves and end point evaluation, Choice and suitability				
	ofindicators in each case.				

Unit III	 Separation Methods: a) Solvent extraction: Efficiency of extraction, Selectivity of extraction, Extraction system, Method of Extraction, applications. b) Solid Phase Extraction: Principle, process and applications. c) Planer Chromatography: Principles, Classification of chromatographic techniques, Techniqueand applications of paper chromatography, Thin–layer chromatography. d) HPTLC conversion of TLC to quantitative measurements, densitometric detectors, fluorimetric detectors. 	15 Lectures
Unit IV	Statistics: Evaluation of Analytical Data, Precision and Accuracy, Types of Errors,Normal Distribution Curve, Standard deviation, Confidence limit. Graphical presentation of result- Method ofaverage, Method of Linear least square,Significant figures. Statistical aid to hypothesis testing: t-test, F-test, chi- 2 test, Correlation.	15 Lectures

1. Skoog D. A., West D. M., Holler and Crouch, Fundamentals of Analytical Chemistry,

Cengage Learning, Wiley-VCH Weinheim, 2011.

2. J. Mendham, R.C.Denney, J.D.Barnes, M.J.K. Thomas, Vogel's Quantitative Chemical Analysis, Pearson Education, ELBS,6th Edition, 2009.

3. Fifield F.W. and Kealey D, Principle & Practice of Analytical Chemistry, Blackwell Science,5th Edition, 2000.

4. Gary D. Christian, Purnendu Dasgupta, Kevin Schug, Analytical Chemistry, John Wiley, 7th Edition, 2013.

5. Douglas A. Skoog, F. James Holler and Stanley R. Crouch, Principles of Instrumental Analysis, Cengage Learning, 6th Edition, 2006.

6. Ahuja & Jespersen, Modern Instrumental Analysis, Elsevier Science, 1st Edition, 2006.

7. D.C. Harris, Exploring Chemical Analysis, W.H. Freeman, 3rd Edition, 2005.

8. Edited by Pradyot Patnaik, Dean's Analytical Chemistry Handbook, McGraw Hill, 2nd Edition, 2004.

9. Klaus Danzer, Analytical Chemistry, Springer-BBH, 2007.

10. A.L. Underwood, Quantitative Analysis, Prentice-Hall of India Pvt Ltd., 1999.

	SEMESTER I	
Code: 1002	Title: Food and Biochemical Analysis	Credits: 4

ii) Able to co iii) Able to p	i) To understand regulation and legislation related to food safety. ompare quality parameters of various food products. perform methods of biochemical analysis.	
-	nt will be able to use methods of food analysis for various products. vledge will enable them to perform better in food industries.	
Paper II		60 Hours
Unit I	a) Regulations & Legislation of Food:	
	Norms, Rules and Regulations, Food Safety and standards Act	15Lectures
	2006, Statistical Process Control, Process Quality and Control	
	Limits.	
	b) Food Additives & Preservatives	
	Characteristics of Food Preservatives, Anti-oxidants, Emulsifiers and stabilizers, Improvers and Bleaching agents, Artificial sweetening agents.	
Unit II	a) Food Quality Parameters:	
	Composition & Impurities in Food, Common Adulterants found in	
	food and Tests. Determination of Moisture, Ash value, Calorific	15Lectures
	value of foods, Saponification value, Acid value, Iodine value,	
	Peroxide value of edible oil.	
	b) Food colors	
	Role of permitted colors, Classification of food colors with	
	chemical structures, Synthetic colors and natural colors,	
	Advantages and Disadvantages of food color.	
Unit III	a) Biochemical analysis: Analysis of blood sample, Serum plasma, Urine test, Blood sugar,	15Lectures
	Electrolytes in body, Substance Glutamic-Oxaloacetic	TUECIOIES
	Transaminase Test (SGOT), Substance Glutamate Pyruvate	
	Transaminase (SGPT), Sputum.	
	b) Body profile:	
	Liver profile, Renal profile, Thyroid profile.	
Unit IV	a) Food analysis: Wheat flour, Milk & Milk products, Tea & Coffee, Honey, Jam &	
	Jelly.	15Lectures
	b) Bread, Biscuits, Ice Cream, Butter, Cheese, Soft drinks,	
	Alcoholic beverages, Cereals and pulses, Confectionery, Fruits,	
	Vegetable, Egg, Fish, Meat.	

1. George Latimer, Official Methods of Analysis of AOAC International (AOAC = Associate of Analytical Communities), Publication – AOAC, 19th Edition,2012.

2. Suzanne Nielson, Food Analysis, Springer, 2010.

3. Yeshajahu Pomeranz, Meloan Editor, Food Analysis Theory & Practice, Springer, 2002.

4. Kirk Sawyer, Pearson Composition & Analysis of Food, Longman Scientific & Technical, 9th Edition, 1992.

5. D. B. Wetzel & G. Charalambous, Instrumental Methods in Food and Beverages Analysis, Elsevier Publication, 1998.

6. M.B. Jacob, Chemical Analysis of Food and Food Products, CBSPB Publisher, 3rd Edition, 2006.

7. M. L. Nollet, Handbook of Food Analysis, CRC Press, 2nd edition, 2004.

8. Semih Otles, Handbook of Food Analysis Instruments, CRC Press, 2008.

9. V. Villaveccha, Treatise on Applied Analytical Chemistry, Methods and Standards for the Chemical Analysis of Industrial and Food Vol I & II, Nabu Press, 2012.

10. Editor Dr. Pico Yolanda, Chemical Analysis of Food: Techniques and Applications, Academic Press, 2012.

11. Semih Otles, Methods of Analysis of Food Components and Additives, CRC Press, 2nd Edition, 2011.

	SEMESTER I	
Code: 1003	Title: Environmental Science	Credits 4
Objectives: i) To ii) To compare (iii) To correlate iv) Acquire know	impact.	
Paper III		60 Hours
UNITI	 a) Air pollution: Sources and sinks of gas pollutants, classification & effects of air pollutants on ecosystem, Air pollution problems in India, Pollution problems in industrial area, Global air pollution problems, Greenhouse effect, Acid rain, Ozone depletion and their consequences on environment, Major air pollution disasters. b) Water pollution: Types of sources and classification of water pollutants, Industrial water pollution, Constituents of aquatic environment, Oxygen contents of water and aquatic life, Oxygen electrode and its use, Mercury pollution and estimation of organomercurials in industrial water, effects of water pollutants on ecosystem. 	15lectures

UNIT II	 a) Methods of control of air pollution: Electrostatic precipitation, wet & dry scrubber, filters, gravity and cyclonic separation, Adsorption, absorption and condensation of gaseous effluent. b) Methods of control of water pollution: water and waste water treatment, aerobic and anaerobic, aeration of water, principle of coagulation, flocculation, softening, disinfection, demineralization and fluoridation. 	15 lectures
UNIT III	 Sampling & analysis of air and water pollutants: a) Methods of sampling gaseous, liquid and solid pollutants, Separation of selected Cations & Anions, Analysis of total cationic and anionic burdens of water. Analysis of gases CO, CO₂, NO₂, SO₂, H₂S.Analysis of toxic heavy metals Cd, Cr, As, Pb, Cu. b) Pesticide: residue analysis, soil pollution, sources of pesticides residue in the environment, pesticides degradation by natural forces, effect of pesticide residue on life, Analytical techniques for pesticides residue analysis. 	15 lectures
UNIT IV	 a) Radiation pollution: Classification & effects of radiation, effects of ionizing radiation on man, effects of non-ionizing radiation on life, radioactivity and nuclear fallout, protection and control from radiation. b) Environmental toxicology: Chemical solutions to environmental problems biodegradability, principles of decomposition for better industrial processes. 	15 lectures

1. A.K. De, Environmental Chemistry, New Age International Publication, 5th Edition, 2005.

2. Wark K. & Werner C., Edited by David and Liptak, Air Pollution, CRC press LLC, 3rd Edition, 2000.

3. S.P. Mahajan, Environmental Pollution Control in Process Industries, Tata McGraw Hill Publishing Co. Ltd, 2008.

4. B. K. Sharma & H. Kaur, Environmental Pollution, Krishna Prakashan Media Pvt Ltd, 2001.

5. R. K. Trivedi, P. K. Goyal, Introduction to Air pollution, ABD publisher, 2003.

6. P. K. Goyal, Water Pollution: Causes Effect and control, New Age International Publication, 2006.

7. S. M. Khopkar, Environmental Pollution Analysis, New Age International Publication, 2011.

8. C. S. Rao, Environmental Pollution Control Engineering, New Age International, 2007.

9. R. Engel, Edited By Ryden Powell, Migula Anderson, Environmental Sciences, Baltic Univ. Publication, 2003.

10. Arvind Kumar, Water Pollution, APH Publishing, 2004.

11. G. S. Sodhi, Fundamental Concepts of Environmental Chemistry, Alpha Science, 2nd Edition, 2005.

12. S. G. Misra & Dinesh Mani, Soil Pollution, APH Publishing Corporation, 2009.

13. S. S. Dara, A Textbook of Environmental Chemistry & Pollution Control, S. Chand Ltd, New Delhi, 2006.

	SEMESTER-I	
Code: 1004	Title: Drug Laws & Packaging	Credits: 4
ii) Able to useiii) To compareiv) To identify	To understand the basic regulation and legislation of drugs. prevention of food adulteration . e standards of ISI and AGMARK. better packaging materials based on advantages and limitations. good practices to become better professional.	
Paper III		60 Hours
Unit I	 a) Legislation and Regulation of Drug: Pharmaceuticals Act, Drugs and Cosmetics Act 1940, US FDA Requirements, EU requirements, International Conference on Harmonization (ICH) guidelines. b) Prevention of Food Adulteration Act & Rules (PFA 1954): Fruit Product Order, Meat Product Order. 	15Lectures
Unit II	 a) Statutory status of pharmacopoeia: Structure of Pharmacopoeia CODEX, Mandatory requirements, Guidelines for license for pharmaceutical industry, Violation of drug laws and statutory punishment. b) Pharmacopoeia: IP, USP, British pharmacopoeia, European pharmacopoeia, National Formulary (NF), Monographs. 	15Lectures
Unit III	 Food Standard Laboratories: The role of Govt. Authorities, their qualification, duties, powers and procedure to be followed. ISI (Indian Standard for Industrial Products), AGMARK (Standardization & grading of Agriculture and allied produce) and other standard for goods & Cosmetic particularly with reference the testing of foods, drug and cosmetic and the raw material concerned. i) Record to be maintained under the Acts ii) Requirements of Quality Control Department for 'WHO' certification, ISO standardization, iii) Implementation process of Accreditation of analytical 	15Lectures

	laboratory.		
Unit IV	a) Packaging materials: Role of packaging material in drug, Food &		
	cosmetic industry, Classification of packaging materials-Glass,	15Lectures	
	Metal, Polymer with their advantages and disadvantages; Testing		
	of material for Packing, Legal consideration in Packing.		
	b) Good Practices: Philosophy of Good manufacturing Practices,		
	Good documentation practices (GDP).		
	Concept of good manufacturing practices (CGMP),		
	Concept of good laboratory practices (CGLP).		

1. Handbook on Modern Packaging Industries, NIIR Project Consultancy Services, Asia Pacific Business Press Inc., 2nd Edition, 2010.

- 2. Edward Baur, Pharmaceutical Packaging Handbook, Publisher Taylor and Francis, 2009.
- 3. G. L. Robertson, Food Packaging Principle & Practice, CRC Press, 3rd Edition, 2012.
- 4. Mehta, Handbook of Drug Laws, University Book Agency Allahabad.
- 5. Govt. of India Publications of Food Drug Cosmetic Acts and Rules.
- 6. Malik Vijay, Laws Relating to Drugs And Cosmetics, Eastern book comp,23rdEdition, 2013.

PRACTICAL ANALYTICAL CHEMISTRY

Objectives: i) Able to prepare standard solutions of various concentrations.

- ii) To develop skills in volumetric titrations.
- iii) Able to separate and estimate elements by solvent extraction method.
- iv) Able to separate and estimate elements and compounds by chromatographic methods.

Code: 1005	PRACTICAL	4 Credits
Volumetric Analysis	Preparation andStandardization of commonly used titrants, Acid-base titration, redox titration, complexometric titration, Precipitation titration, Non-aqueous titrations.	8 Hours/week
Solvent		
Extraction&	Separation and estimation of elements	
Chromatographic		
Method		

PRACTICAL FOOD AND BIOCHEMICAL ANALYSIS

Objectives: i) Able to analyze milk and milk products.

- ii) To compare analysis of tea and coffee.
- iii) To develop skills in analytical methods of food products

Code: 1006	PRACTICAL	4 Credits
Food Analysis	Milk and Milk Products, Tea, Coffee, Honey, Preservatives. Jam, Jelly, Squash, Edible Oil, Pickle, Sauce, Vinegar	8 Hours/week

SEMESTER II

Code: 2001	Title: Electro Analytical and Spectroscopic Methods	Credits: 4
Objectives	; i) To compare basic concepts of electro analytical and spectroscopic metho	ds.
ii) Able to o	orrelate principle and working of different types of instruments used for an	alysis.
iii) Able to	use these techniques in research and analysis.	
iv) To apply	<i>i</i> these techniques in the work place.	
Paper I		60 Hours
Unit I	 Electro Analytical Methods-I a) Potentiometry: Electrodes, electrode systems, Determination of cell Electro Motive Force (EMF), Potentiometric titrations. b) Ion Selective electrodes: solid state, precipitate, liquid-liquid, Enzyme and Gas sensing electrodes with applications, Ion selective field effect transistors, Bio-catalytic membrane electrodes, Enzyme based biosensors. 	15 Lectures
Unit II	 Electro analytical methods-II a) Polarography: Basic principles, Diffusion current, polarized electrode, Microelectrode, Dropping Mercury Electrode, Ilkovic equation, Polarographic wave, Qualitative, and Quantitative analysis. b) Stripping Methods of analysis. c) Coulometry at controlled potential, coulometric titration 	15 Lectures
Unit III	 Absorption Spectroscopic methods a) UV-Visible molecular Absorption Spectroscopy: Instrumentation, Absorbing species, Qualitative and Quantitativeanalysis. b) Atomic absorption Spectrometry: Theory of atomic absorption spectrometry, Instrumentation, interferences. c) Molecular IR absorption Spectroscopy: Theory of Infrared Absorption, Spectrometry, Instrumentation, Sample handling, Qualitative and Quantitative analysis, Fourier Transform Infrared Spectrometry (FTIR). 	15 Lectures

Unit IV	Emission Spectroscopic methods	15 Lectures
	a) Molecular Fluorescence spectroscopy: Theory of fluorescence and	
	phosphorescence, variable that affectsfluorescence and	
	phosphorescence. Instruments for measuring fluorescence,	
	phosphorescence, application of fluorescence and phosphorescence.	
	b) Flame emission spectroscopy: Instrumentation, Flame	
	characteristics, Flame processes, EmissionSpectra, Quantitative	
	measurements, Interferences, Applications.	
	c) Turbidimetry and Nephlometry: Theory, Instrumentation,	
	Applications.	

1. Skoog D. A., West D.M., Holler and Crouch, Fundamentals of Analytical Chemistry,

Cengage Learning, Wiley-VCH Weinheim, 2011.

2. J. Mendham, R. C. Denney, J. D. Barnes, M. J. K. Thomas, Vogel's Quantitative Chemical Analysis, Pearson Education, ELBS, 6th Edition, 2009.

3. Fifield F. W. and Kealey D, Principle & Practice of Analytical Chemistry, Blackwell Science, 5th Edition, 2000.

4. Gary D. Christian, Purnendu Dasgupta, Kevin Schug, Analytical Chemistry, John Wiley, 7th Edition, 2013.

5. Douglas A. Skoog, F. James Holler and Stanley R. Crouch, Principles of Instrumental Analysis, Cengage Learning, 6th Edition, 2006.

6. Ahuja& Jespersen, Modern Instrumental Analysis, Elsevier Science, 1st Edition, 2006.

7. D.C. Harris, Exploring Chemical Analysis, W.H. Freeman, 3rd Edition, 2005.

8. Edited by Pradyot Patnaik, Dean's Analytical Chemistry Handbook, McGraw Hill, 2nd Edition, 2004.

9. Klaus Danzer, Analytical Chemistry, Springer-BBH, 2007.

10. A.L. Underwood, Quantitative Analysis, Prentice-Hall of India Pvt Ltd., 1999.

SEMESTER-II			
Code: 2002	Title: Pharmaceutical Analysis	Credits: 4	
Objectives: i) To identify active drug ingredients in drug products.		
ii) To underst	and dosage form and its mode of administration.		
iii) Able to re	fer and compare pharmacopoeias for different parameters and analysis.		
iv) To correla	te basic concept of QA and QC in pharma industries.		
Paper II		60 Hours	
UNIT I	Introduction:	15 Lectures	
	a) Active Pharmaceutical Ingredients (API) and drug products.		
	b) Dosage form: Brief information on manufacturing and		
	administration-Tablets, Capsules, Injections, Ointments, Creams, Oral		
	solution, Aerosol.		
	c) Control release formulation.		

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UNIT II	a) Introduction to Pharmacopoeia and its importance.	15 Lectures
	b) Application of analytical techniques in pharmaceutical industries	
	- Official sources of Impurities and limit test (As, Pb, Fe, Chloride,	
	Sulphate).	
	- Techniques of Analysis: Introduction, physical tests, chemical	
	tests, conventional test, Gravimetric, Volumetric.	
	- Instrumental Techniques : UV-Visible, IR, Flame photometry,	
	Chromatography (TLC,HPLC, GC).	
UNIT III	a) Assay of main classes of drugs Chemotherapeutic agents:	15 Lectures
•••••	Introduction, Type, Properties, Method of Analysis.	
	b) Dissolution and Disintegration, Assay of drugs, Vitamin C, Vitamin	
	B ₂ , Aspirin, Ibuprofen, Streptomycin sulphate, Calcium lactate, Iron	
	tablets (Autrin, Fefol capsule), Laxative substances, Anta-acid.	
UNIT IV	a) Quality Assurance (QA)	15 Lectures
	- Concept of Total Quality Management,	
	- Documentation and its importance in QA.	
	b)Quality Control (QC)	
	- Change control management, Out of specifications (OOS), Deviation	
	reporting, Stability studies (QA Pharma), Quality control, laboratory	
	responsibilities, routine controls, Calibration of instruments,	
	Standard test procedures.	
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Beckett & Stenlake, Practical Pharmaceutical Chemistry Vol (I) (II), CBS Publisher, 4thEdition, 2006.
 V. K. Selvaraj, Practical Pharmaceutical Chemistry, Campus Books International Publisher, 2012.

Editor J. D. Nally, Good Manufacturing Practices for Pharmaceuticals, CRC Press, 6th Edition, 2006.
 Edited by A. A. Signore and T. Jacobs, Good Design Practices for GMP Pharmaceutical Facilities, CRC Press, 2005.

5. Indian Pharmacopeia (Latest Edition)

6. British Pharmacopeia (Latest Edition)

7. Edited by Allen, Loyd Jr., Remington The Science & Practice of Pharmacy, Publication - Pharmaceutical Press, 22nd Edition,2012.

8. D.G. Watson, Pharmaceutical Analysis, Churchill Livingstone Publisher, 3rdEdition, 2012.

9. Edited by L. Ohannesian and A. Streeter, Handbook of Pharmaceutical Analysis, CRC Press, 2001.

10. M. E. Aulton, Dosage Form, Churchill Livingstone Publisher, 2nd Edition, 2001.

- 11. J. Wiley, Text Books of pharmaceutical Analysis, Connors, 2001
- 12. Higuchi, Chemical Analysis of Drug, Interscience, 1995.

SEMESTER II		
Code: 2003	Title: Cosmetics Formulation & Quality Control	Credits: 4
Objectives: i) Enable to understand processes of cosmetic formulations.	
ii) To compre	hend advantages and limitations of raw materials used in cosmetic form	ulations.
iii) Able to as	sess importance of quality control process in cosmetic industries.	
iv) Able to pe	erform analysis of cosmetic formulations for professional growth.	
Paper IV		60 Hours
Unit I	a) Introduction, Common processes used for cosmetic	15 Lectures
	formulations, raw materials used in cosmetics.	
	b) Colors: Natural and synthetic, their advantages & limitations.	
	c) Perfumes: Natural and synthetic, their advantages & limitations.	
Unit II	a)Cosmetic formulation:	15 Lectures
	Introduction, Characteristics, BIS standards of: Skin creams, Face	
	Powders, Lipsticks, Shampoos, Manicure preparations, Hair	
	grooming preparations, Nail lacquers, Dentifrices.	
Unit III	a) Herbal Cosmetics products for: Skin, Nails, hairs, dentifrices.	15 Lectures
	b) Test methods for cosmetic products: Contact urticaria, Primary &	
	secondary irritants, Skin sensitivity, Patch, Photo-patch, Repeated	
	insult etc.	
Unit IV	a) Quality control of:	15 Lectures
	Cosmetics raw materials- Goods inspection report, determination	
	of specific microorganisms (Escherichia, salmonella, pseudomonas,	
	staphylococcus etc) and total viable aerobic count (membrane	
	filtration, plate count, serial dilution);	
	Intermediate and bulk finished products–Fineness, texture,	
	apparent density, color shade & match, odor, uniformity of	
	emulsion, softening point, melting range, foam and foam stability.	
	b) Analysis of cosmetics: Face powder (fats and fatty acids, boric	
	acid, zinc, total titanium & iron), Lipstick (separation of waxes and	
	oil & analysis of dyes), Nail enamel (Bismuth Oxy Chloride, free	
	formaldehyde), Shampoo (analysis of nonvolatile matter, borate,	
	sulphate, phosphates, surfactants), Creams (types of emulsion, %	
	water, ash & chloroform soluble material).	

1. Edited by Martin M. Reiger, Harry's Cosmeticology, Chemical Publishing Co. Inc,8th Edition, 2009.

2. P.P. Sharma, Cosmetics, Formulations, Manufacturing and Quality Control, Vandana Publication Ltd,4th Edition,2010.

3. M. S. Balsam & E. D. Sagarin, Cosmetics Science & Technology, Wiley Interscience Publication, 2nd Edition, 2008.

4. H. Panda, Herbal Cosmetics, Asia Pacific Business Press Inc., 2008.

5. B. M. Mittal & R. N. Saha, Handbook of Cosmetics, VallabhPrakashan, New Delhi, 2008.

6. S. Nanda and R. K. Khar, Cosmetic Technology, Birla Publications Pvt Ltd, 1stEdition, 2006.

7. George Latimer, Official Methods of Analysis of AOAC International (AOAC = Associate of Analytical Communities), 19th Edition, Publication – AOAC, 2012.

8. M. L. Schlossman, Chemistry and Manufacture of Cosmetics, Allured Publishing Corporation,4th Edition, 2009.

9. A. O. Barel, N. Paye, H. I. Maibach, A Handbook of Cosmetics Science and Technology, 3rdEdition, 2009.

A. Salvador, A. Chisvert, Analysis of Cosmetic Products, Publisher – Elsevier, 2011.
 P. Elsner, H. I. Maibach, Cosmeceuticals and Active Cosmetics, Taylor & Francis, 2nd Edition, 2005.

Code: 2004		Title: Research Methodology	Credits: 4
Objectives: i) F	Participants will be at	ble to work more efficiently with knowledge of b	basic principles.
ii) Able to colle	ect data, literature su	rvey for research project.	
iii) Able to use	statistical tests for a	nalysis and presentation of data.	
iv) Participant	will be better equipp	ed with the knowledge of chemical safety and c	lisaster
management t	o work in research fi	eld/industries.	
Paper IV			60Hours
Unit I	safety, Working with Principles ofsolution b) The investigative SI units (Internationa method and design of	pratory Techniques: Basic principles, Health and a liquids, Basic laboratory procedures I & II, chemistry, pH and buffer solutions. approach: Making and recording measurements, al System of Units) and theiruse, Scientific of experiments, Project work.Collection of Data), literature survey & review.	15 lectures
Unit II	in tables, Hints forso Choosing and using	entation of data: Using graphs, Presenting data olving numerical problems, Descriptive statistics, statistical tests,drawing chemical structures, oputational chemistry.	15 lectures

	b) Statistical Packages for Social Science (SPSS) Workshop.	
Unit III	a) Chemical safety: General safety and operational rules, Safety equipments, Personal protective equipments, Compressed gas safety, Safety practices for disposal ofbroken glassware,Chemicals, Centrifuge safety, Treated biomedical wastes and scientific ethics.	15 lectures
Unit IV	Disaster Management: Emergency response to: Chemical spills, Radiation spills, Biohazard spills, Leakingcompressed gas cylinders, Fires, Medical emergency, Accident reporting (Bhopal gas tragedy, Chernobyl three mile island).	15 lectures

1. J. R. Dean, A. M. Jones, D. Holmes, R. Reed, J. Weyersand A Jones, Practical Skills in Chemistry, Pearson Education Ltd. [Prentice Hall], 2002.

2. C. R. Kothari, Research Methodology: Methods and Techniques, New Age International, 2013.

3. A. K. Singh, Tests, Measurements and Research Methods in Behavioral Sciences, Bharti Bhawan Publisher And Distributor, 2012.

4. Martyn Denscombe, The Good Research Guide, McGraw-Hill International, 2007.

5. Ranjit Kumar, Research Methodology, Sage Publication Ltd, 3rd Edition, 2011.

6. Edited by J.C. Taylor, Advances in Chemistry Research, Vol 17, Nova Science Publishers INC, 2013.

7. Oklahoma State University Laboratory Safety Manual, 1999.

8. Margaret Diane LeCompte, Wendy L. Millroy, Judith Preissle, The Handbook of Qualitative Research in Education, Academic Press Inc

9. Bohdan O, Szuprouiez, Multimedia Networking, Mcgraw-Hill

10. Introduction to Research, Tynes Hillway Houghton Wiffin Company, 2005.

PRACTICAL SPECTROSCOPY AND CHROMATOGRAPHY

Objectives: i) Able to handle colorimetric instrument for analysis.

ii) To develop skills in chromatographic techniques for analysis.

Code: 2005	PRACTICAL	4 Credits
Spectroscopy	Colorimetric analysis of elements, Mixture, Simultaneous estimation of metals, pk value of indicator by Spectrophotometry.	8 Hours/week
Chromatography	Chromatography- Ion- exchangechromatography, Thin layer chromatography.	

PRACTICAL PHARMACEUTICAL ANALYSIS

Objectives: i) Able to analyze various drugs by standard methods. ii) Able to compare dissolution and disintegration test for different drugs.

Code: 2006	PRACTICAL	4 Credits
Pharmaceutical Analysis	Assay of alkaloids, Vitamins, Antibiotics, Sulpha drugs, Anta-acids, Anti-bacterials.	8 Hours/week
	Dissolution test, Disintegration test, Weight variation test, Test for uniformity of content.	

SEMESTER III			
Code: 3001		Title: Advanced Chromatography and Spectroscopic Methods	Credits: 4
ii) To correlate	principle	ehend basic concept of chromatographic and spectroscopic r and instrumentation of various instruments used. omatographic and spectroscopic methods.	nethods.
Paper I			60 Hours
Unit I	Introdu Gas chr Chroma Size Exc exclusic	Chromatography I ction to column chromatography omatography: Principle of GLC, Instruments for GLC, Gas atographic Columns and Stationary phases, Applications. clusion Chromatography: Column packing, Theory of size of on chromatography, Application of size exclusion tography.	15 lectures
Unit II	HPLC: P Ion Chro packing Super C Instrum	chromatography II rinciple and Instrumentation. omatography: Ion exchange equilibria, Ion-exchange 's, Applications of Ion Chromatography ritical Fluid Chromatography:Properties of SFC- ientation and operatingvariables, Comparison with other f chromatography, Applications.	15 lectures

Unit III	Advanced Spectroscopic Methods I	15 lectures
	Mass spectrometry: Molecular Mass Spectra, Ion sources,	
	Analyzers, Detectors, Mass Spectrometers, Application of	
	Molecular Mass Spectrometry, Secondary Ion Mass Spectrometry.	
	Atomic Emission Spectroscopy Inductively Coupled Plasma: Principle, Instrumentations and applications. Nuclear Magnetic Resonance: Theory of NMR-Quantum description, Classical description of NMR, Relaxation Processes in NMR. Environmental effects on NMR Spectra-Chemical shift, spin splitting, Rules governing the interpretation of first order spectra, effect of chemical exchange, NMR Spectrometers, Applications of proton NMR, C-13 NMR.	
Unit IV	Advanced Spectroscopic Methods II	15 Lectures
	Raman spectroscopy: Theory of Raman spectroscopy,	
	Instrumentation, Application.	
	Hyphenated Methods:	
	Gas Chromatography – Mass Spectrometry (GC-MS),	
	Gas Chromatography – IR Spectrometry (GC-IR),	
	Liquid Chromatography – Mass Spectrometry (LC-MS),	
	Tandem Mass Spectrometry (MS-MS),	
	Inductively Coupled Plasma – Mass Spectrometry (ICP-MS).	

1. Skoog D. A., West D. M., Holler and Crouch, Fundamentals of Analytical Chemistry,

Cengage Learning, Wiley-VCH Weinheim, 2011.

2. J. Mendham, R. C. Denney, J. D. Barnes, M. J. K. Thomas, Vogel's Quantitative Chemical Analysis, Pearson Education, ELBS, 6th Edition, 2009.

3. Fifield F. W. and Kealey D, Principle & Practice of Analytical Chemistry, Blackwell Science, 5th Edition, 2000.

4. Gary D. Christian, Purnendu Dasgupta, Kevin Schug, Analytical Chemistry, John Wiley, 7th Edition, 2013.

5. Douglas A. Skoog, F. James Holler and Stanley R. Crouch, Principles of Instrumental Analysis, Cengage Learning, 6th Edition, 2006.

6. Ahuja & Jespersen, Modern Instrumental Analysis, Elsevier Science, 1st Edition, 2006.

7. D.C. Harris, Exploring Chemical Analysis, W.H. Freeman, 3rd Edition, 2005.

8. Edited by Pradyot Patnaik, Dean's Analytical Chemistry Handbook, McGraw Hill, 2nd Edition, 2004.

9. Klaus Danzer, Analytical Chemistry, Springer-BBH, 2007.

10. A. L. Underwood, Quantitative Analysis, Prentice-Hall of India Pvt Ltd., 1999.

	SEMESTER III	
Code: 3002	Title: Organic Analysis	Credits: 4
Objectives:) To comprehend spectroscopic methods for characterization	of organic compounds.
II) Able to ide	entify spectra for structure elucidation.	
III) To compil	e implication of significance and application of carbon Nanotu	bes.
IV) Able to co	orrelate importance of organic synthesis.	
Paper II		60 Hours
Unit I	a) UV-visible Spectroscopy: Introduction, Electronic	15 Lectures
	transitions, Applications, Calculation of λ max.	
	b) IR-Spectroscopy: Basic concept, Fundamental	
	vibrations, Functional group frequencies, Identification of	
	organic compounds.	
Unit II	a) HNMR: Use of HNMR for characterization and	15 Lectures
	structural elucidation of organic compounds and	
	interpretation of spectra.	
	b) CNMR: Use of CNMR for characterization and structural	
	elucidation of organic compounds and interpretation of	
	spectra.	
	c)Mass Spectroscopy: Ionization on electron impact, Mass	
	Spectrum, detection of isotopes, Fragmentation and	
	spectra of few classes of compounds.	
Unit III	a) Functional group analysis: Introduction, principles &	
	methods of analysis.	15 Lectures
	b) Nanotechnology: Introduction; Carbon nanotubes:	15 Lectures
	Significance, Preparative methods, Types SWNT, MWNT	
	and applications. Nanomaterials.	
Unit IV	a) Organic synthesis: Principles of organic synthesis,	15 Lectures
	Wacker process, Ziegler Natta, Grignard Reaction,	
	Williamson Synthesis.	
	b) Organic trace analysis: Introduction, Units, Sampling,	
	Concentration techniques and estimation methods.	
	c) Micro-elemental analysis of C,H,N,O and halogens.	

1. Donald L. Pavia, Jeorge S. Kriz, James R. Vyvyan, Spectroscopy by Lampmen, Cengage Learning India Pvt Ltd, 4th Edition, 2012.

2. R. A. Sheldon, Isabella Arends, ULF Hanefeld, Green Chemistry & Catalyst, Wiley VCH Verlag GmBH and Company, 2007.

3. J.H. Clark, D.J. Macquarrie, Handbook of Green Chemistry and Technology, John Wiley & Sons, 2008.

4. Sulbha K. Kulkarni, Nanotechnology Principles and Practices, Capital Publishing Company, 2007.

5. A. I. Vogel, Elementary Practical Organic Chemistry: Small Scale Preparations Part I, Dorlin Kindersley India Pvt. Ltd., 2011.

6. Edited by M. B. Smith, March's Advance Organic Chemistry, John Wiley & Sons, 7th Edition, 2013.

7. P. S. Kalsi, Spectroscopy of Organic Compounds, New Age International, 6th Edition, 2004.

8. Mzike Lancaster, Green Chemistry an Introductory Text, Royal Society of Chemistry, 2002.

9. R. M. Silversteime and G. C. Bassler, Spectrometric Identification of Organic Compounds, Johan Wiley Publication, 1991.

10. S. Siggia and J.G. Hanna, Quantitative Organic Analysis Via Functional Groups, Wiley Interscience, 1979.

11. Harald Gunther, NMR Spectroscopy, Wiley –VCH Weinheim, 2013.

SEMESTER III				
Code: 300)3	Title: Microbiological Methods of Analysis	Credits: 4	
-	• •	croorganism based on their morphology. e medium for different microorganisms.	<u> </u>	
		ods of staining for microorganism. I symptoms of food, water, air borne diseases.		
•		ition for air, water and food borne diseases.		
Paper III			60 Hours	
Unit I	General Charact	of Microorganisms: teristics of Algae, Protozoa, Mycoses, Pathogens, s, Fungi, Yeasts and Parasites.	15 Lectures	
	 b) Staining method: Gram staining, Acid-base staining, Staining of spores, Staining of bacterial spores. 			
Unit II		viruses, Classification, Replication cycle, Effects of viruses Viruses of medical importance, Rubella viruses infection,	15 Lectures	

	Laboratory test for Polioviruses, Lentiviruses and AIDS (Acquired	
	Immunodeficiency Syndrome), Detection tests for HIV (Human Immunodeficiency Virus).	
	b) Bacteria:	
	Introduction to bacteriology, Basic features of bacteria, Classification,	
	Shape and structure of bacteria, Factors affecting growth of bacteria.	
Unit III	a) Culturing of Microorganism	
	Introduction, Composition and types of culture media, Preparation of culture media, Quality control of various culture media. b) Control of microorganism	15 Lectures
	Control of microorganisms by chemical agents, Characteristics of an ideal chemical agents, Major groups of disinfectants and antiseptics, Evaluation of antimicrobial potency of disinfectants and antiseptics.	
Unit IV	a) Food borne diseases:	15 Lectures
	Bacteria responsible for food borne diseases, Signs and Symptoms of bacterial infection, Antibacterial drugs.	15 Lectures
	b) Water-borne diseases:	
	Microbiology of drinking water, Microorganism responsible for water	
	borne diseases, Signs & Symptoms	
	c) Airborne diseases:	
	Air borne diseases and microorganism, Signs& symptoms	

1. Godkar P. S. and Godkar D. P., Textbook of Medical Laboratory Technology, Bhilani Publishing House, 2nd Edition, Mumbai, 2006.

2. Chatterjea M. N. & Shinde R., Textbook of Medical Biochemistry, J. P. Brothers Medical Publishers PVT, New Delhi, 8th Edition, 2012.

3. Jindal Meenakshi, Introduction to Waste Water Treatment Process, Mangalam Publication, Delhi, 1st Edition, 2009.

4. Jay James, Modern Food Microbiology, CBS Publisher & Distributor, 2005.

5. K. Vijya Ramesh, Food Microbiology, MJP Publisher, Chennai, 2007.

6. Pelczar M., Chan E. C. S., Krieg Noel R., Microbiology, Tata McGraw Hill, New Delhi, 2003.

7. Doyle M., Beuchat L., Food Microbiology, Fundamentals and Frontiers, Montville ASM Press Washington D. C., 2013.

8. Prescott, Harley and Klein, Textbook of Microbiology, Mc-Graw Hill,7th Edition, 2008.

9. A. Anantnarayan and C. K. Panikar, Textbook of Microbiology, Orient Longman Pvt.Ltd., 7th Edition, 2005.

SEMESTER III			
Code: 3004A		ELECTIVE I: Industrial Product and Forensic Analysis	Credits: 4
Objectives: i) T	o familia	rize importance of industrial products.	
ii) To compare	content	analysis of different types of steel, cements and ceramics	5.
iii) To acquire l	knowledg	ge of surfactants as pollutant.	
iv) To compreh	end soil	and fertilizer analysis.	
v) To compare	forensic	analysis of blood and hair.	
Paper IV			60 Hours
Unit I	Analysis	s of Steel & Cement	15 Lectures
	a) Analy	sis of Steel and Ferrous materials: Sampling, Analysis of	
	steel ar	d ferrous alloy: Carbon, Silicon, Manganese,	
	phosph	orous, sulphur, nickel, chromium, vanadium, tungsten,	
	molybd	enum, cobalt, titanium.	
	b) Analy	ysis of Cement and building materials: Type of cement,	
	Samplir	ng, Analysis of Cement and building materials: Silicon	
	dioxide	, aluminium oxide, ferric oxide, calcium oxide,	
	magnes	ium oxide, sulphur trioxide, sulphide-sulphur, loss on	
	ignition	, insoluble residue, sodium and potassium oxide.	
Unit II	Analysi	s of Ceramic materials:	15 Lectures
	a) Analy	sis of quartz: Volatile residue, Zirconium dioxide,	
		ium oxide, Calcium and magnesium oxides, Sodium and um oxide.	
	-	ysis of Clays and feldspars: Determination of moisture,	
		dioxide, Total oxides, Ferric oxides, Titanium dioxide,	
		um oxide, Calcium oxide, Magnesium oxide.	
	c) Analy and lea	vsis of glasses: Types of glasses, determination of lead	
	d) Analy	ysis of ceramics: Determination of Titanium dioxides,	
		minium oxide from oxide ceramics.	
		ysis of ferrous slags: Determination of iron, calcium and sium, total oxides.	

Unit III	 Analysis of other products a) Surfactants and Detergents: Introduction to different pollutants, Classification, Representative methods of analysis. b) Effluent water analysis: Introduction to different pollutants, Classification, Representative methods of analysis. 	15 Lectures
	c) Soil and fertilizer analysis: Characteristic of different soils, Classification, Representative methods of analysis of soil parameters, pH texture, Water retaining capacity, Sodium Absorption Ratio (SAR) value, NPK value of fertilizers.	
Unit IV	 Forensic Science a) Forensic Analysis: Overview, destructive & nondestructive techniques, data interpretation. b) Blood Analysis: Blood preservation and ageing effect, analysis of blood components & exogenic substances, blood stain analysis. c) Hair analysis: Structure and composition of hair, Morphological examination, Chemical analysis of hair components & components remaining on or in hair. 	15 Lectures

1. F. J. Welcher, Standard Methods of Chemical Analysis, Vol. 2, (Part A & B), Von Nostrand & Rober E. Krieger Publishing Co. New York, 5th Edition, 2000.

2. Edited by J. Mendham, R. C. Denney, M. Thomas, B. Sivasankar, Vogel's Quantitative Chemical Analysis, Dorlin Kindersley (India) Pvt. Ltd, 6th Edition, 2009.

3. William D. Callister Jr., D. G. Rethwisch, Fundamentals of Materials Science & Engineering, John Wiley and Sons. Inc., New York, 4th Edition, 2012.

4. Suzanne Bell, Forensic Chemistry, Pearson Education Publishers, 2nd Edition, 2012.

5. Lwrance Kobilinsky, Forensic Chemistry Handbook, Wiley Interscience, 2011.

6. J. I. Khan, T. J. Kenedy, D. R. Christian, Basic Principles of Forensic Chemistry, Humana Press, 2011.

7. Robert A. Meyers, Encyclopedia of Analytical Chemistry, Wiley Interscience, 2012.

8. V. Raghavan, Material Science & Engineering, PHI Learning Pvt Ltd Publisher, 5th Edition, 2004.

9. Thomas M. Schmitt, Analysis of Surfactants, Marcel Dekker Inc, 2nd Edition, 2001.

SEMESTER III			
Code: 3004B	ELECTIVE II: Medicinal Chemistry	Credits: 4	
Objectives: i) T	Objectives: i) To familiarize students with various types of medicines.		
ii) Able to comp	ii) Able to compare mode of administration and bioavailability of drugs.		

· ·	are drugs acting on CNS.	
	e drugs action and toxic effects of steroids.	60 Hours
Paper IV UNIT I	 General consideration of following drugs: The development of the following classes of drugs including structure activity relationship, mode of action and mechanism of action of drugs in that class, mode of administration, bioavailability and its relation to structure, chemical nomenclature, generic names and toxic effects. 1. Antiseptic and disinfectants 2. Chemotherapeutic agents; thermotherapy of acid fast infection (Anti-tubercular and anti-leprotic agents) 3. Chemotherapeutic agents of parasitic infection, antimalarials, where the test is a structure in the structure is a structure of the structure is a structure of the s	60 Hours 15 Lectures
	 anti-amoebic, anti-trypanosomiasis and antihelmintic agents 4. Antifungal agents 5. Anti-viral agents 6. Anti-neoplastic agents. 	
UNIT II	 Antibiotics Sulphonamides Diuretics Hypoglycemic agents Diagnostic agents and pharmaceutical aids Miscellaneous drugs like anticoagulants and antilipemic agents. 	15 Lectures
UNIT III	 Drugs acting on central nervous system : 1. General anesthetics, hypnotics and sedative anti-convulsants 2. Analgesics: narcotic agents 3. Anti-pyretic, anti-rheumatics and anti-inflammatory agents 4. CNS Stimulants 5. Psychotherapeutics agents; major and minor tranquilizers, anti-depressant 6. Local anesthetics 	15 Lectures
UNIT IV	 Drugs acting on nervous system: 1. Cholinergic stimulants and blocking agents 2. Adrenergic stimulants and blocking agents 3. Hypotensive agents and cardiovascular drugs 4. Vitamins and co-enzyme 5. Antihistamines 6. Corticosteroids, sex steroids, oral contraceptives and synthetic estrogens. 	15 Lectures

1. Edited by Donald J. Abraham, Burger's Medicinal Chemistry and Drug Discovery, 7th Edition, Wiley Publication, 2010.

2. Thomas L. Lemke, David A. Williams, V. Roche & S.W. Zito, Foye's Principles of Medicinal Chemistry, Published by Lippincott Williams & Wilkins, 7th Edition, 2012.

3. Gareth Thomas, Medicinal Chemistry, Wiley Interscience Publication, 2nd Edition, 2011.

4. V. Alagarsamy, Textbook of Medicinal Chemistry, Reed Elsevier India Pvt. Ltd, 2010.

5. G. P. Elllis & G. B. West, Progress in Medicinal Chemistry, Vol 9, North Holland Publishing Co., 2011.

6. D. Sriram & P. Yogeeswari, Medicinal Chemistry, Publisher Dorling Kindersley (India) Pvt. Ltd, 2008.

7. Grahm L. Patrik, An Introduction to Medicinal Chemistry, Oxford University Press, 5th Edition, 2013.

8. Ashutosh Kar, Medicinal Chemistry, Lippincott Williams & Wilkins, 2007.

9. Hugo Kubinyi, QSAR: Hansch Analysis and Related Approaches, Wiley Interscience Publication, 2008.

10. Dr. S. S. Kadam, Principles of Medicinal Chemistry, Pragati Books Pvt Ltd, 2008.

SEMESTER III				
Code: 3004C	ELECTIVE III: Biosensors, Agrochemicals & Organic Polymers	Credits: 4		
Objectives: i) To understand basic concept of biosensors and their applications.				
ii) To identify d	ifferent types of agrochemicals and their analysis.			
iii) To study syn	thesis and analysis of organic polymers .			
Paper IV		60 Hours		
UNIT I	Biosensors:	15 Lectures		
	Definition, components and types of Biosensors.			
	Enzyme based Biosensors.			
	Immobilization of Enzymes.			
	Types of Transducers, fabrication of probes for sensors and			
	evaluation of sensor performance.			
	Application of Biosensors.			
UNIT II	Analysis of Agrochemicals:	15 Lectures		
	Introduction, Classification, Mechanism of action and synthesis.			
	Insecticides: DDT (Dichloro Diphenyl Trichloro Ethane), BHC			
	(Benzene Hexa Chloride), Aldrin, Endosulfon, Malathion,			
	parathion.			
	Herbicides: 2,4-dichloro phenoxy acetic acid, Dalapon, Paraquat,			
	Banalin, Butacarb.			
	Fungicides: Bordeaux mixture, Copper oxychloride, Zineb, Benomyl			
	(Benlate).			
	Analysis of pesticides residue and toxicological effects.			
UNIT III	Organic Polymers:	15 Lectures		
	Basic Concepts, Polymers, monomers, molecular weight and molar			

	 mass, end groups, degree of polymerization. Nomenclature Classification of polymer-thermoplastic, thermosetting. Copolymers – random, alternate, block and graft copolymers, Molecular architecture, Polymerization and functionality. Polymerization processes – addition and step polymerization. Mechanism of polymerization – free radical and ionic. Heterogeneous polymerization – Zeigler-Natta catalysis. Optical Activity in polymers, stereo isomers-isotactic, atactic and syndiotactic polymers. 	
UNIT IV	Analysis of Polymers: Molecular weight of polymers - Arithmetic weight average and number average molecular weights of polymers. Determination of molecular weight of polymers, End-Group analysis, Cryoscopy, Light scattering, Viscosity, and gel permeation chromatographic methods. Thermal Transition in Polymers, Tg and Tm and their relation, Differential Scanning Calorimetry (DSC), Thermal Gravimetric Analysis (TGA), Spectral method of analysis.	15 Lectures

1. V. C. Yang, T. T. Ngo, Biosensors and Their Applications, Kluver Academic Publisher Newyork, 2000.

2. Rajmohan Joshi, Biosensors, Isha Books India, 2006.

3. J. J. Murphy, Handbook of Residue Analytical Methods for Agrochemicals by Wiley Interscience Publication, 2003.

4. Edited by Q. A. Acton, Agrochemicals: Advances in Research & Application, Scholarly Editions Book, 2012.

5. W. Brown, C. Foote, I. Iverson, E. Anslyn, Organic Chemistry (Organic Polymers), Cengage Learning Publisher, 2008.

6. F. W. Bilmayer, Text Book of Polymer Science, Wiley India Pvt. Ltd., 2008.

7. V. K. Ahluwalia, A. Mishra, Polymer Science A Text Book, Taylor and Francis Group Publisher, 2008.

8. Manas Chanda, Introduction to Polymer Science and Chemistry, CRC Press (Taylor and Francis Group), 2nd Edition, 2013.

9. S. J. Knowles, Quantitative Analysis of Agrochemicals by Spectroscopy Technique, University of Reading Publisher, 1996.

10. A. Townshed, Encyclopedia in Analytical Science, Academic Press, 1995.

PRACTICAL ADVANCED SPECTROSCOPIC METHODS

Objectives: i) To compare different methods of spectroscopic analysis.

ii) To develop skills in flame photometry and fluorimetry instruments.

Code: 3005	PRACTICAL	4 Credits
Advanced spectral methods UV spectroscopy	Spectroscopic determination of elements, Standard addition method and method of least squares, extractive photometry, photometrictitration.	8 hours/week
Fluorimetry	Fluorometric determinations of organic compounds by Calibration curve, standard addition method.	
Flame photometry	Flame photometric determination of alkali metals by calibration curve method.	

PRACTICAL ORGANIC ANALYSIS

Objectives: i) To develop skill in identification of organic compounds on the basis of their spectra.

ii) Able to estimate organic compounds on the basis of functional groups.

Code: 3006	PRACTICAL	4 Credits
Identification &Estimation of Functional Groups	Identification: Alcoholic, phenolic carbonyl, carboxylic, ester, nitro, amino group, amide group, degree of unsaturation, hydrocarbons, olefins using sample spectra. Estimation: Amines, phenols,aldehydes,ketones, Ester, amide, Carboxylic compounds.	8 hours/week

SEMESTER IV		
Code: 4001	Title: Advanced Analytical Techniques	Credits: 4
Objectives: i) To understand basic concepts of advanced analytical techniques. ii) To comprehend principle and instrumentation of advanced analytical techniques. iii) To understand role of computers in analytical chemistry. iv) To apply the concepts of green chemistry to analytical chemistry for better environment.		
Paper I		60 Hours

Unit I	 Advanced Electro-analytical methods a) Amperometric Titrations: Principle, Instrumentation, titration curves. BiamperometricTitrations: Principle, instrumentation, titration curves. b) Modified Polarographic Methods: Differential Pulsed polarography,Square Wave Polarography,Cyclic voltammetry. 	15 Lectures
Unit II	 Thermal & Radioactive methods of analysis a) Thermal methods: TGA :- Principles, Instrumentation, Application DTA :- Principles, Instrumentation, Application DSC :- Principles, Instrumentation, Application Thermometric Titrations b) Radio analytical Methods: Radioactive decay products and processes, Radioactive decay rates, Instrumentation. Neutron activation method:- Destructive, non-destructive. Applications of Neutron Activation. c) Isotope dilution method:Principle, Applications. Photo Acoustic Spectroscopy (PAS): Photo acoustic effect, theory of PAS, Instruments, Applications. 	15 Lectures
Unit III	 Surface Analytical Techniques a) Surface spectroscopic methods: Electron Spectroscopy for Chemical Analysis (ESCA) - Principles, Instrumentations, Applications. b) Auger electron spectroscopy: Principles, Instrumentations, Applications. c) Scanning Electron Microscopy: Principles, Instrumentation, Applications. d) Computers in Analytical chemistry: Computer terminology, Components of computer, Computer software-Scientific packages, Applications of computers- Passive andActive, Computers networks, Lab Information Management System (LIMS). 	15 Lectures
Unit IV	Green Analytical Methods: Principle & concept, Atom economy, and trends in Green Analytical chemistry, Greening sample treatment, Green Instrumental Analysis.	15 Lectures

PRACTICAL ADVANCED ANALYTICAL TECHNIQUES

Objectives: i) To develop skill in conductometry, pHmetry and thermometry for analysis of acids and bases.

ii) Able to analyze different parameters for water analysis.

iii) To analyze cosmetic raw materials.

Code: 4002	PRACTICAL	4 Credits
Conductometry	Conductometric titration of acids, bases, mixture of	8 hours (usal
pHmetry	acids. pH metric titration of mixture of acids and selection of indicators for volumetric titration, pH metric titration of polybasic acids.	8 hours/week
Thermometric titrations	Thermometric titrations of acids/bases	
Water analysis	Water Analysis of samples for hardness, dissolved oxygen, residual chlorine content,	
Cosmetic analysis	Analysis of Cosmetics intermediates.	

Reference Books:

1. Skoog D.A., West D.M., Holler and Crouch, Fundamentals of Analytical Chemistry,

Cengage Learning, Wiley-VCH Weinheim, 2011.

2. J. Mendham, R. C.Denney, J. D. Barnes, M. J. K. Thomas, Vogel's Quantitative Chemical Analysis, Pearson Education, ELBS, 6th Edition, 2009.

3.Fifield F. W. and Kealey D, Principle & Practice of Analytical Chemistry, Blackwell Science,5th Edition, 2000.

4. Gary D. Christian, Purnendu Dasgupta, Kevin Schug, Analytical Chemistry, John Wiley, 7th Edition, 2013.

5. Douglas A. Skoog, F. James Holler and Stanley R. Crouch, Principles of Instrumental Analysis, Cengage Learning, 6th Edition, 2006.

6. Ahuja & Jespersen, Modern Instrumental Analysis, Elsevier Science, 1st Edition, 2006.

7. D.C. Harris, Exploring Chemical Analysis, W.H. Freeman, 3rd Edition, 2005.

8. Edited by Pradyot Patnaik, Dean's Analytical Chemistry Handbook, McGraw Hill, 2nd Edition, 2004.

9. Klaus Danzer, Analytical Chemistry, Springer-BBH, 2007.

10. A. L. Underwood, Quantitative Analysis, Prentice-Hall of India Pvt Ltd., 1999.

11. M. Koel & M. Kaljurand, Green Analytical Chemistry, RSC Publishing, 2012.

12. M. L. Guardia, S. Carrigues, A Handbook of Green Analytical Chemistry, Wiley Interscience, 2012.

Code: 4003

Objectives: i) Able to collect data and literature survey.

ii) Able to use statistical aids for data processing.

iii) Acquire knowledge and skills for higher level research work.

Research Project & Evaluation:

Students will be given guidance on basic research approaches, methodology and tools; and expected to collect data (primary, secondary), conduct literature survey and complete a small research project in the institution or industries.

After completion of research study, each student will submit the report of 50-60 pages (12,000 to 15,000 words). The chapter wise indexing and typing rules will be the same as of thesis writing.

Each report will be evaluated by the guide and external referee. Evaluation of research project will be conducted internally and externally.

Code: 4004	IN-PLANT TRAINING	8 Credits (6 wks)
		• •

Objectives: i) The intern will develop skills in the analytical techniques in practical work situation.

ii) Intern will develop and strengthen their professional skills and interpersonal relationship in work environment of the industry.

iii) It will provide opportunity to find out their interest in a particular carrier.

iv) Experiential learning will have broader perspective for employment.

In-plant Training & Evaluation:

Internship placement will be in research institutes, industries like pharma, food and cosmetic or Government organization (BARC or Haffkine Institute), subject to approval of the internship coordinator. Student will prepare a detailed plan with the approval of external mentor which will be submitted to the internship coordinator within 7 days of the joining. At the end of internship, student will prepare a write up (2000 to 3000 words) based on their daily work activities in the organization. Evaluation of the internship assignment will be done internally and externally.

First to Fourth Semester

Question Paper Pattern for Theory

Time:	3 Hrs	Marks: 100 (Converted to 50 Marks)
	Note: All Questions are Compulsory	
Q.1	Attempt any TEN of the following Objective	Ques 20 Marks
	15 questions from Unit I, II, I	II & IV.
Q.2	Attempt any TWO of the following	20 Marks
	Four questions from Unit I.	
Q.3	Attempt any TWO of the following	20 Marks
	Four questionsfrom Unit II.	
Q.4	Attempt any TWO of the following	20 Marks
	Four questionsfrom Unit III.	
Q.5	Attempt any TWO of the following	20 Marks

Four questions from Unit IV.

First to Fourth Semester

Question Paper Pattern for Practical(internal/external)

Lab Course		Time: 4 Hrs.
		Marks: 50
Q.1	Synopsis of Experiment	05Marks
Q.2	Experiment	30 Marks
Q.3	Viva	10 Marks
Q.4	Record Book	05 Marks